

**Introduction to General Chemistry CHM 1025C**  
**Ch 4-5-11 Notes**

**Instructor: Mr. George W.J. Kenney, Jr**

**Sodium Chloride** \_\_\_\_\_

**Carbon Monoxide** \_\_\_\_\_

**Potassium Iodide** \_\_\_\_\_

**Nitrogen Dioxide** \_\_\_\_\_

**Calcium Sulfide** \_\_\_\_\_

**Selenium Hexafluoride** \_\_\_\_\_

**Cesium Bromide** \_\_\_\_\_

**Silicon Dioxide** \_\_\_\_\_

**Magnesium Oxide** \_\_\_\_\_

**Cobalt (III) Chloride** \_\_\_\_\_

**Copper (I) Iodide** \_\_\_\_\_

**Tin (IV) Bromide** \_\_\_\_\_

**Mercury (II) Chloride** \_\_\_\_\_

**Lead (II) Sulfide** \_\_\_\_\_

**Potassium Nitride** \_\_\_\_\_

**Mercury (II) Oxide** \_\_\_\_\_

**Rubidium Fluoride** \_\_\_\_\_

**Sodium Hydride** \_\_\_\_\_

**Chromium (II) Floride** \_\_\_\_\_

**Magnesium Bromide** \_\_\_\_\_

**Manganese (II) iodide** \_\_\_\_\_

**Lithium Oxide** \_\_\_\_\_

**Diiodine Heptoxide** \_\_\_\_\_

**Carbon Dioxide** \_\_\_\_\_

**Carbon Tetrafluoride** \_\_\_\_\_

**Ammonia** \_\_\_\_\_

**Hosphorous Trichloride** \_\_\_\_\_

**NaCl**

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**KI**

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**CaS**

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**CsBr**

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**MgO**

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**AlCl<sub>3</sub>**

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**MgI<sub>2</sub>**

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**CoCl<sub>3</sub>**

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**CuI**

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**SnBr<sub>4</sub>**

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**HgCl<sub>2</sub>**

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**PbS**

---

**K<sub>3</sub>N**

---

**HgO**

---

**RbF**

---

**NaH**

---

**CrF<sub>2</sub>**

---

**MgBr<sub>2</sub>**

---

**MnI<sub>2</sub>**

---

**Li<sub>2</sub>O**

---

**I<sub>2</sub>O<sub>7</sub>**

---

**CO<sub>2</sub>**

---

**CF<sub>4</sub>**

---

**NH<sub>3</sub>**

---

**PCl<sub>3</sub>**

---

**CO**

---

**NO<sub>2</sub>**

---

**SeF<sub>6</sub>**

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**SiO<sub>2</sub>**

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